

Structures and Energetics of $E_2H_3^+$ (E = As, Sb, and Bi) Cations

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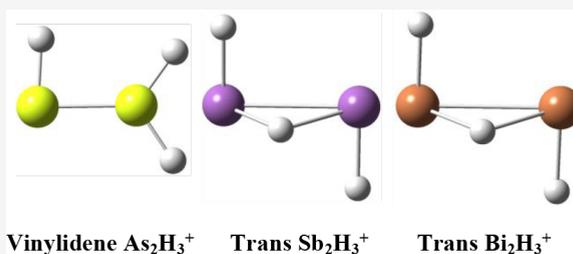


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ABSTRACT: E_2H_2 (E = As, Sb, Bi) structures involving multiple bonds have attracted much attention recently. The $E_2H_3^+$ cations (protonated E_2H_2) are predicted to be viable with substantial proton affinities (>180 kcal/mol). Herein, the bonding characters and energetics of a number of $E_2H_3^+$ isomers are explored through CCSD(T) and DFT methods. For the $As_2H_3^+$ system, the CCSD(T)/cc-pVQZ-PP method predicts that the vinylidene-like structure lies lowest in energy, with the trans and cis isomers higher by 6.7 and 9.3 kcal/mol, respectively. However, for $Sb_2H_3^+$ and $Bi_2H_3^+$ systems, the trans isomer is the global minimum, while the energies of the cis and vinylidene-like structures are higher, respectively, by 2.0 and 2.4 kcal/mol for $Sb_2H_3^+$ and 1.6 and 15.0 kcal/mol for $Bi_2H_3^+$. Thus, the vinylidene-like structure is the lowest energy for the arsenic system but only a transition state of the bismuth system. With permanent dipole moments, all minima may be observable in microwave experiments. Besides, we have also obtained transition states and planar-cis structures with higher energies. The current results should provide new insights into the various isomers and provide a number of predictions for future experiments.



INTRODUCTION

Main-group chemistry involving multiple bonds between heavier elements is a rapidly developing field.^{1–7} Particularly, the compounds of group 15 elements, such as the dipnictogen $HEEH$ (E = N, P, As, Sb, Bi) molecules, have been studied systematically.^{8,9} Among these, in comparison with the diazenes and diphosphenes, the heavier congeners diarsenes, distibenes, and dibismuthenes are less frequently reported, showing the difficulties in stabilizing multiple bonds between the heavier atoms. In 1997, Tokitoh, Arai, Okazaki, and Nagase synthesized the first stable dibismuthene, in which a Bi=Bi double bond was characterized by means of UV–vis and Raman spectra, X-ray crystallographic structural analysis, and computations.¹⁰ In 1998, the same research group synthesized the stable distibene ($RSb=SbR$).^{11,12} Later, Schulz et al. determined the solid-state structures of Et_4Sb_2 and Et_4Bi_2 using X-ray diffraction.¹³ In 2015, Scheer and co-workers reported the isolation and structural authentication of $HAsAsH$ in a bulky diuranium(IV) complex.¹⁴ The same group in 2019 incorporated the $HAs=AsH$ moiety as a side-on-coordinated ligand in a simple mononuclear $Fe(CO)_4$ complex.¹⁵

Theoretically, as early as 1990, Nagase, Suzuki, and Kurakake predicted that the Sb and Bi atoms can form double-bonded compounds at the HF/DZ(d,p) level of theory.¹⁶ In 2008 and 2014, Su et al. explored the lowest singlet and triplet potential energy surfaces for the $HAsXH$ and $HSbXH$ (X = N, P, As, Sb, Bi) systems with the QCISD/LANL2DZdp method, and their results are in good agreement with the limited experimental results.^{17,18} In 2020, Li, Huang,

Xie, Robinson, and Schaefer investigated the E_2H_2 (E = As, Sb, Bi) molecules with the CCSD(T) method as well as four different density functional theory (DFT) methods.¹⁹ They found that the trans isomer lies lowest in energy, but both trans and cis structures may be observable due to large barriers between them. Their natural bond orbital (NBO) analyses showed that, as expected, the Wiberg bond indices (WBIs) for the E=E bond in the trans, cis, and vinylidene-like structures are all close to 2.0.

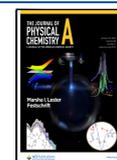
Related viable cations are appealing synthetic targets. It is common knowledge that for alkenes the C=C double bond can be protonated to form carbocations. In 2016, the protonation of disilicon(0) compounds was reported by Filippou et al.,²⁰ and a topomerization was found between the “ σ -bonded” protonation tautomer and the “ π -bonded” disilahydronium ion. For the group 15 elements, the methylation of a diphosphene ($Mes^*P=PMe^*$) was reported to form a stable phosphanyl phosphonium cation.²¹ However, the corresponding protonation of the diphosphene was not observed.²² To our knowledge, the cations designed by adding a proton to the E_2H_2 (E = As, Sb, Bi) molecules have not been reported. For these reasons, these cations are worthy of study.

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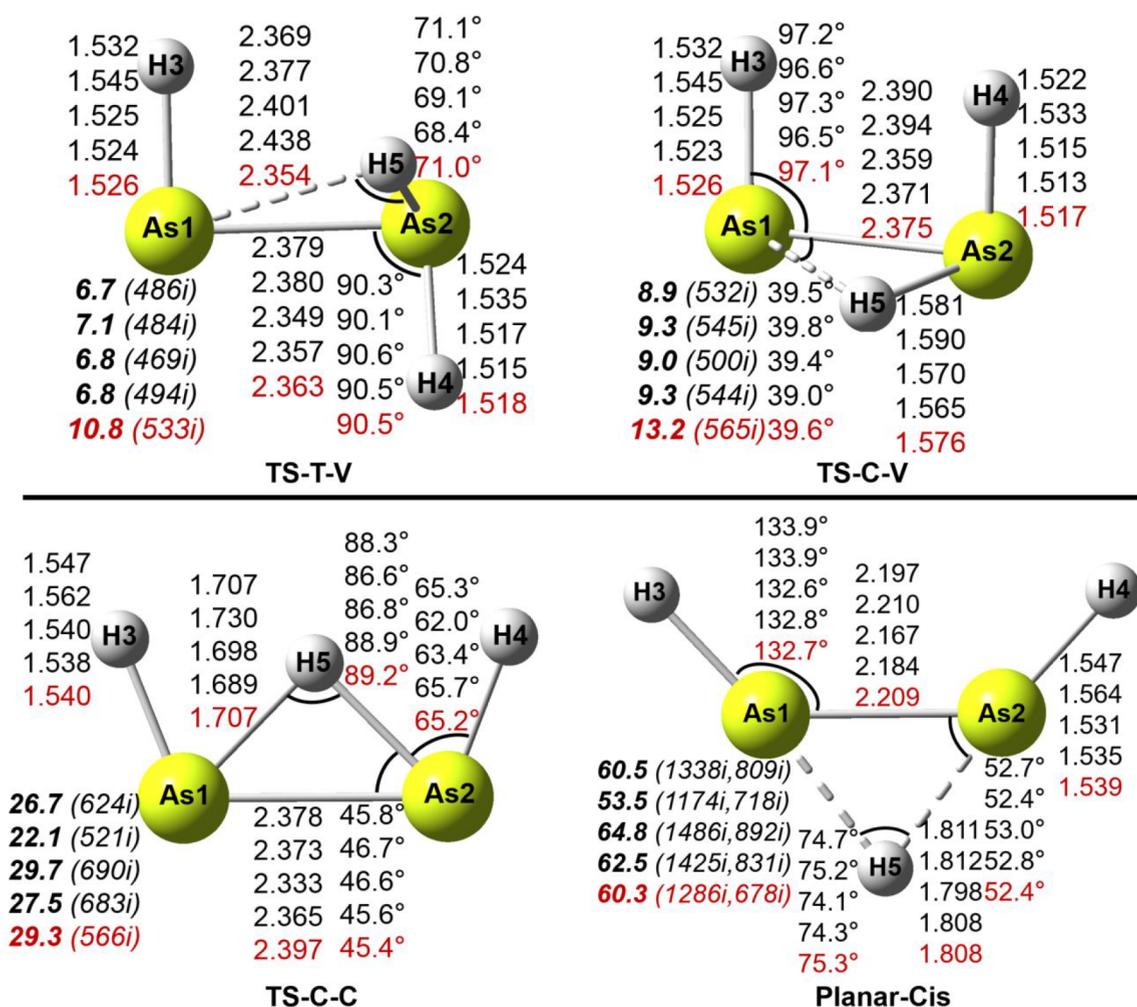


Figure 2. Geometries and energetics for four As_2H_3^+ transition states. Bond distances are in Å and relative energies (**bold face**) to the vinylidene-like global minimum (in Figure 1) in kcal/mol. The imaginary vibrational frequencies (in cm^{-1}) are shown in parentheses.

planar-cis (Figures 1 and 2). The vinylidene structure is predicted to be lowest-lying, different from the energy order for neutral As_2H_2 , for which the global minimum is the trans structure.^{15,19} The trans As_2H_3^+ structure is predicted to lie 6.7 kcal/mol higher than the vinylidene structure by the CCSD(T)/cc-pVQZ-PP method, and the cis structure lies 9.3 kcal/mol above the vinylidene structure. Interestingly, the energy order of the As_2H_3^+ isomers (vinylidene and trans) is similar to that for the observed protonated cation $[\text{Si}_2(\text{H})(\text{Idipp})_2]^+$ of the disilicon(0) compound.²⁰ The latter most favored structure is the Si–H “ σ -bonded” minimum, while the “ π -bonded” minimum is higher in energy by 7.3 kcal/mol. A transition state between them has a relative energy of 10.3 kcal/mol.²⁰

For the TS-T-V transition state, the IRC analysis shows that it connects the trans and vinylidene structures with an energy barrier of 10.8 kcal/mol with respect to the vinylidene side and 4.1 kcal/mol from the trans side at the CCSD(T)/cc-pVQZ-PP level. The TS-C-V structure is a transition state connecting the cis and vinylidene structures with a barrier height of 3.9 kcal/mol from the cis minimum and 13.2 kcal/mol from the vinylidene-like side. The TS-C-C transition state is found to connect the two mirror images of the cis isomer with an energy barrier of 20.0 kcal/mol. The planar-cis structure has two

imaginary vibrational frequencies with a very high relative energy (60.3 kcal/mol) above the vinylidene minimum.

Figure 1 shows that the DFT-optimized geometries for all As_2H_3^+ structures are in qualitative agreement with the CCSD(T) results. The global-minimum vinylidene-like structure is planar, and its As–As bond distance is 2.235 Å (with the CCSD(T)/cc-pVQZ-PP method), which is 0.08 Å longer than that (2.154 Å) in the neutral vinylidene-like As_2H_2 .¹⁹ The As–As bond distances in the trans and cis structures are predicted to be 2.361 and 2.377 Å, longer than that of the vinylidene-like structure by 0.13 and 0.14 Å, respectively. The CCSD(T)-predicted harmonic vibrational frequencies for As–As stretching are 259 (vinylidene), 321 (trans), and 312 (cis) cm^{-1} , respectively (Table S3). The As–As distances for the transition states TS-C-C, TS-C-V, and TS-C-C are 2.363, 2.375, and 2.397 Å, respectively. The As–As bond distance in the planar-cis structure is the shortest, i.e., 2.209 Å, suggesting the strongest As=As double bond.

According to the NBO analyses, the As atoms in all As_2H_3^+ isomers display positive charges, and the sum of charges on the two As atoms is close to +1 (Table 1), indicating that the added charge resides primarily in the vicinity of the two As atoms. The trans and cis structures are “ π -bonded” isomers, which makes the proton attract electron density from the As–As π -bonding orbital, leading to a lower As–As bond order.

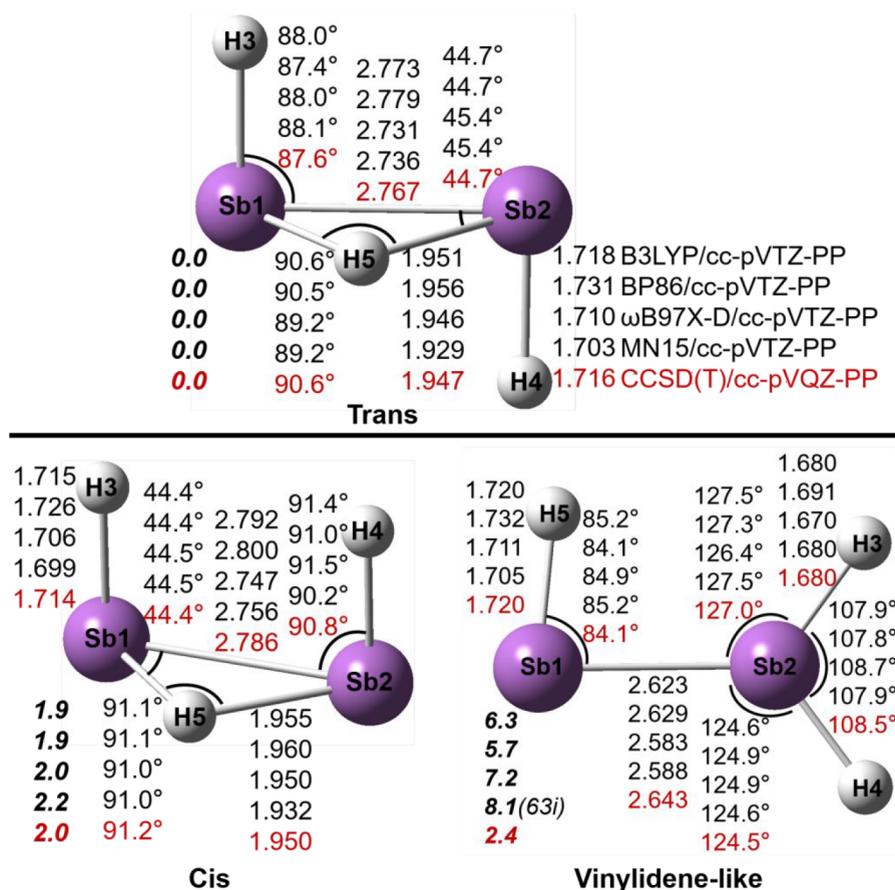


Figure 3. Geometries and energetics for three Sb_2H_3^+ equilibrium geometries. Bond distances are in Å and energies (**bold face**) in kcal/mol.

Table 1 shows that the WBI values for the As–As bonds for the As_2H_3^+ trans and cis isomers are 1.28 and 1.25, respectively, much lower than their corresponding WBI values for the As_2H_2 trans (2.03) and cis (2.01) isomers.¹⁹ The vinylidene structure is a “ σ -bonded” isomer, which makes the proton attract electron density from an As lone pair orbital, with little effect on the As=As double bond. Thus, the corresponding WBI value is 1.90 (Table 1), comparable with the WBI value (1.99) for the neutral As_2H_2 vinylidene isomers.¹⁹ Obviously, these As–As WBI values are consistent with the As–As bond distances (Table 1). Accordingly the As–As WBI values for the transition states are 1.37 (TS-T-V), 1.33 (TS-C-V), and 1.52 (TS-C-C), while that for planar-cis is 2.08 (Table 1).

Sb_2H_3^+ Structures. We have found eight structures for the Sb_2H_3^+ system (Figures 3 and 4 and Table 2). Like As_2H_3^+ in the previous section, the trans, cis and vinylidene-like structures are predicted to be genuine minima. However, unlike As_2H_3^+ the lowest-lying Sb_2H_3^+ structure is the trans structure with the added proton “ π -coordinated”. The cis and vinylidene Sb_2H_3^+ structures lie above the trans structure by 2.0 and 2.4 kcal/mol, respectively. For Sb_2H_3^+ , a staggered minimum is also found, but only by the DFT methods, lying 7.8 kcal/mol above the trans structure at the MN15/cc-pVTZ-PP level. However, with the CCSD(T)/cc-pVQZ-PP method, the staggered structure is a transition state connecting the trans and vinylidene minima. The DFT methods located a transition state TS-T-S connecting the trans and staggered structures with only a tiny energy barrier from the staggered side (<0.2 kcal/mol with the ω B97X-D and MN15 methods). Another

transition state, TS-C-V, is found between the cis and vinylidene minima, lying 5.2 kcal/mol in energy above vinylidene and 5.6 kcal/mol above cis. The TS-C-C structure is a transition state connecting two mirror cis minima with a high barrier (19.0 kcal/mol). The planar-cis structure is a second-order stationary point (with two imaginary vibrational frequencies, 1072i and 586i cm^{-1}), and it lies 56.9 kcal/mol above the trans structure, comparable to that (60.3 kcal/mol) for the planar-cis structure of the analogous As_2H_3^+ system.

The Sb–Sb bond distances in the trans and cis Sb_2H_3^+ structures are predicted to be 2.767 and 2.786 Å, respectively, longer than those in the corresponding neutral Sb_2H_2 structures by 0.14 Å.¹⁹ Accordingly, the Sb–Sb WBI values decrease from ~ 2.0 for the neutral Sb_2H_2 molecules to ~ 1.2 for the Sb_2H_3^+ cations (Table 2). The decrease of the Sb–Sb bond order for the trans and cis structures arises because the added proton breaks the Sb–Sb π bond to form an Sb–H–Sb two-electron three-center ($2e-3c$) bond. The Sb–Sb bond distance in the vinylidene-like structure (2.643 Å) is shorter, and its WBI value is larger (1.84) (Table 2), since the added proton approaches a lone pair orbital, leaving the Sb–Sb π bond little affected. The planar-cis structure has the shortest Sb–Sb bond distance (2.609 Å), consistent with its largest WBI value (2.01) and its largest Sb–Sb stretching vibrational frequency (227 cm^{-1} ; Table S3), similar to the planar-cis structure of As_2H_3^+ . The transition states were found to have slightly longer Sb–Sb distances than their related minima, i.e., TS-T-S (2.791 Å), TS-C-V (2.800 Å), and TS-C-C (2.874 Å).

Bi_2H_3^+ Structures. The geometries of the Bi_2H_3^+ stationary points are reported in Figures 5 and 6. Similar to Sb_2H_3^+ , the

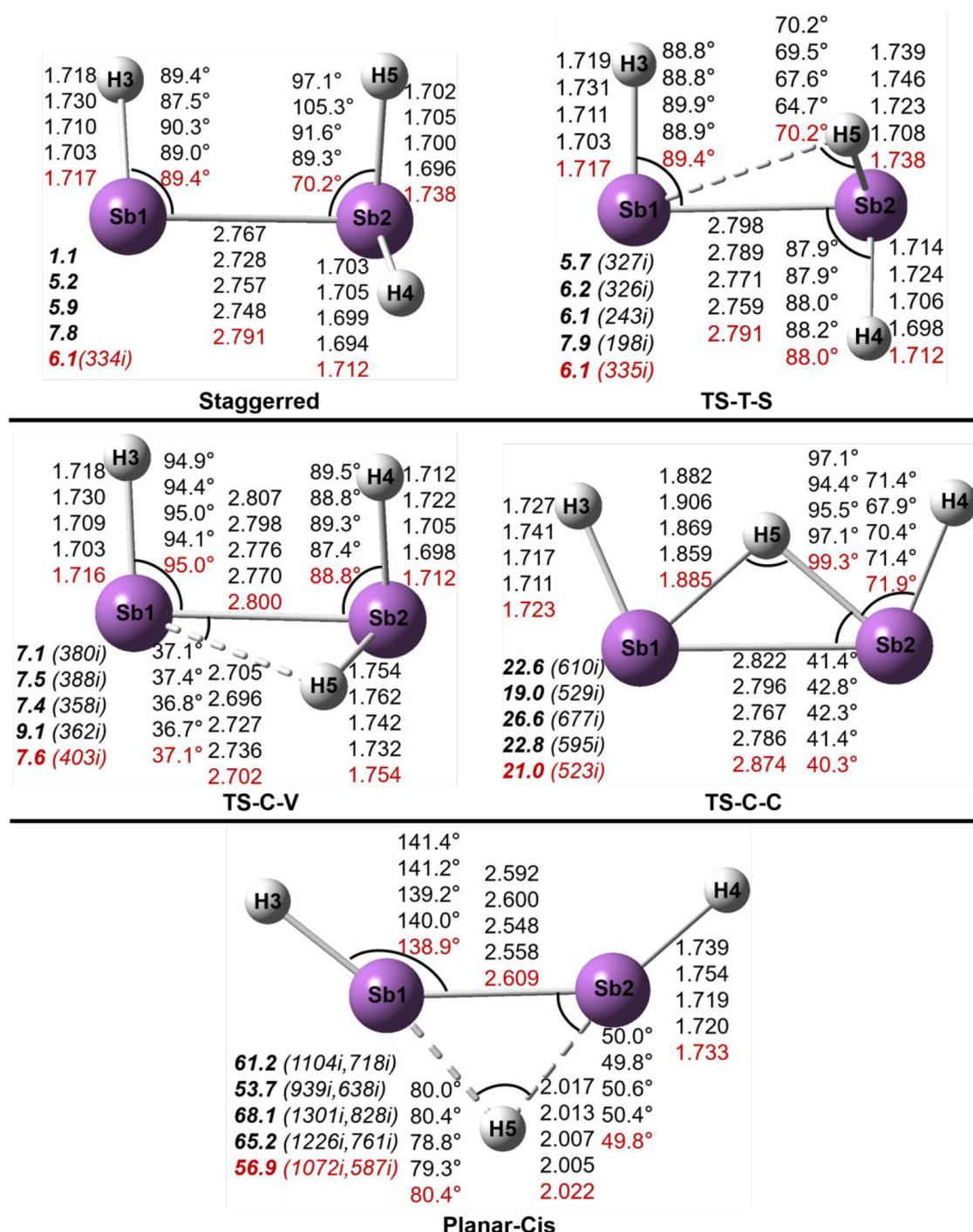


Figure 4. Geometries and energetics for five Sb_2H_3^+ transition states. Bond distances are in Å and energies (**bold face**) relative to the trans global minimum (in Figure 3) in kcal/mol. The imaginary vibrational frequencies (in cm^{-1}) are shown in parentheses.

trans structure is the global minimum, while the cis structure is 1.6 kcal/mol above trans. Unlike the As and Sb analogues, the vinylidene-like structure for Bi_2H_3^+ is a transition state predicted by all theoretical methods and lies 15.0 kcal/mol above trans with the CCSD(T) method. The normal mode corresponding to the imaginary vibrational frequency is the $-\text{BiH}_2$ wag. The IRC analysis shows that the vinylidene transition state connects two mirror trans structures. Structure TS-T-C is a transition state between the trans and cis structures, lying 9.2 kcal/mol above trans. The TS-C-C structure, similar to those for Sb_2H_3^+ and As_2H_3^+ , is a

transition state connecting two mirror cis structures with the energy barrier of 16.8 kcal/mol, comparable to 19.0 kcal/mol for Sb_2H_3^+ and 20.0 kcal/mol for As_2H_3^+ . Interestingly, we found another transition state, TS'-C-C, that also connects two mirror cis isomers with a lower energy barrier (10.6 kcal/mol). Unlike the analogous As_2H_3^+ and Sb_2H_3^+ systems, there is no stationary planar-cis saddle point for the Bi system, since it collapses to TS-C-C.

The Bi–Bi bond distance in the trans Bi_2H_3^+ isomer is predicted to be 2.945 Å (Figure 3 and Table 3), much longer than that for neutral Bi_2H_2 , 2.780 Å (theoretical)¹⁹ or 2.821

Table 2. Relative Energies (ΔE , in kcal/mol) and Sb–Sb Bond Distances (R , in Å) for Sb_2H_3^+ at the CCSD(T)/cc-pVQZ-PP level of theory and Sb–Sb Wiberg Bond Indices (WBIs) and Sb Atomic Charges (Q_{Sb}) from Natural Bond Orbital (NBO) Analysis at the MN15/cc-pVTZ-PP Level

structure	ΔE	$R_{\text{Sb-sb}}$	$\text{WBI}_{\text{Sb-sb}}$	Q_{Sb}
trans	0.0	2.767	1.21	0.70/0.70
cis	2.0	2.786	1.19	0.69/0.69
vinylidene	2.4	2.643	1.84	0.63/0.46
staggered ^a	7.8	2.748	1.34	0.85/0.37
TS-T-S	6.1	2.791	1.29	0.87/0.39
TS-C-V	7.6	2.800	1.24	0.87/0.43
TS-C-C	21.0	2.874	1.36	0.63/0.63
planar-cis	56.9	2.609	2.01	0.52/0.52

^aThe values for the staggered structure are from the MN15 method, since it is not a minimum with the CCSD(T) method.

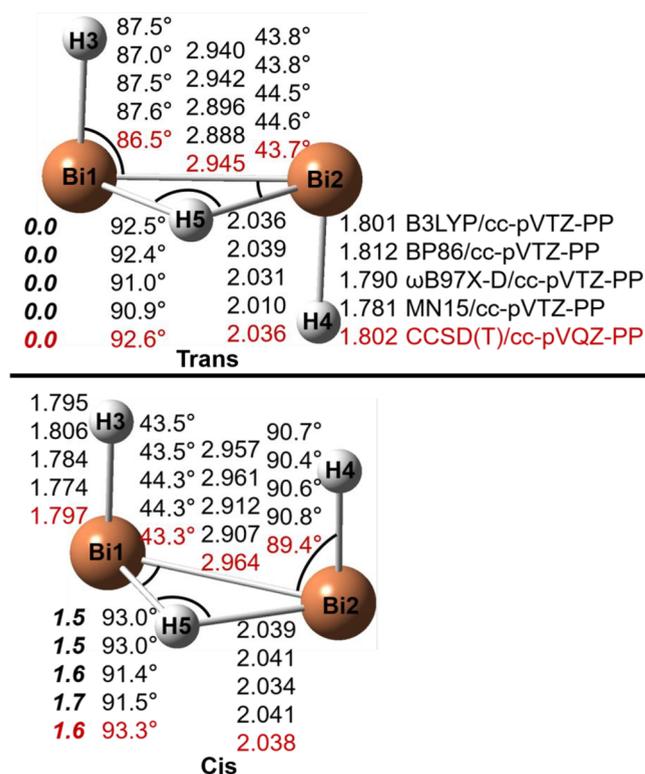


Figure 5. Geometries and energetics for two Bi_2H_3^+ equilibrium geometries. Bond distances are in Å and energies (bold face) in kcal/mol.

and 2.854 Å (experimental values from crystal structures).^{13,31} Similar to the case of trans As_2H_3^+ and Sb_2H_3^+ , the added proton weakens the Bi–Bi π bond and decreases the Bi–Bi WBI value from 2.02 (neutral Bi_2H_2) to 1.20 (cationic Bi_2H_3^+). Similarly, the Bi–Bi distance in the cis Bi_2H_3^+ isomer is predicted to be 2.964 Å, longer than that for the neutral Bi_2H_2 (2.794 Å).¹⁹ The Bi–Bi WBI value is 1.18, smaller than that (2.01) for neutral Bi_2H_2 .¹⁹ The vinylidene structure has a shorter Bi–Bi bond distance (2.814 Å) and larger WBI value (1.71), while the three transition states have longer Bi–Bi distances and smaller WBIs (Table 3).

DISCUSSION

Geometries and Energetics. Compared with previous theoretical investigations on the neutral E_2H_2 molecules of group 15 ($\text{E} = \text{As}, \text{Sb}, \text{Bi}$),¹⁹ we find some similarities and some differences for the E_2H_3^+ cations upon the addition of H^+ to the neutral E_2H_2 molecules. For example, we find the trans and cis structures to be genuine minima for all three E_2H_3^+ systems, with the cis structures lying always higher than the trans structures by ~ 2 kcal/mol. This is similar to the situation for the neutral E_2H_2 molecules.¹⁹ However, with an added “ σ -coordinated” E–H bond, the vinylidene-like structure is a global minimum for As_2H_3^+ , lying below the trans structure by ~ 7 kcal/mol, while the vinylidene-like structure for Sb_2H_3^+ is still a minimum but lies above the trans structure with a “ π -coordinated” E–H bond by ~ 2 kcal/mol. Finally, the vinylidene-like structure for Bi_2H_3^+ is a transition state lying above the trans structure by ~ 15 kcal/mol (Table 4). This should be attributed to the weaker Sb–Sb and Bi–Bi π bonds compared to the As–As π bond, following the periodic trend. A diagram of MOs (Figure 7) for the E–E π bonds ($\text{E} = \text{As}, \text{Sb}, \text{Bi}$) shows such a trend.

For Sb_2H_3^+ , a staggered minimum was located by the four DFT methods, but it seems to be not viable since there is a very small energy barrier (< 0.2 kcal/mol predicted by MN15 and ω B97X-D) to depart to the trans structure. In fact, the high-level CCSD(T) method predicts that no staggered minimum exists, but instead a staggered transition state directly falling to the trans minimum. A planar-cis structure is found for the As_2H_3^+ and Sb_2H_3^+ systems. However, this structure should have little chemical significance because it is a second-order saddle point and has very high energy (> 50 kcal/mol above the global minimum).

We have also tried to optimize C_{2v} structures between two mirror vinylidene structures using the DFT and CCSD(T) methods. However, these C_{2v} structures are either second-order saddle points with very high energies (76, 76, and 97 kcal/mol above vinylidene-like for $\text{E} = \text{As}, \text{Sb},$ and Bi , respectively) or lead to dissociation, depending on the electron configurations. Thus, we do not discuss these C_{2v} structures in the text but report the high-lying geometries in the Supporting Information.

Proton Affinity. The proton affinity (PA) is an important property for molecules and atoms and is related to the basicity in the gas phase. The study of proton affinities will provide useful information with respect to structure, stability, and bonding. Using the results for E_2H_3^+ ($\text{E} = \text{As}, \text{Sb}, \text{Bi}$) in the present paper and the E_2H_2 results in a previous paper,¹⁹ we can directly obtain the proton affinities for E_2H_2 . At the CCSD(T)/cc-pVQZ-PP level of theory, the adiabatic proton affinities are predicted to be 180 kcal/mol for As_2H_2 , 186 kcal/mol for Sb_2H_2 , and 192 kcal/mol for Bi_2H_2 (Table 5). No experimental results are currently available for these species, and our predictions provide new data for them. As a comparison, these values are close to the experimental PA (192 kcal/mol) for the related molecule N_2H_2 .³⁷ The substantial PA values indicate that the added proton strongly stabilizes the neutral E_2H_2 species.

The NBO analyses may help understand the PA values for E_2H_2 ($\text{E} = \text{As}, \text{Sb}, \text{Bi}$). For example, from the NBO analysis, one resonance structure of the trans E_2H_3^+ ($\text{E} = \text{As}, \text{Sb}, \text{Bi}$) species has an occupied E1–H5 bond orbital and an empty lone pair orbital on the E2 ($\text{E} = \text{As}, \text{Sb}, \text{Bi}$) atom, where H5 is

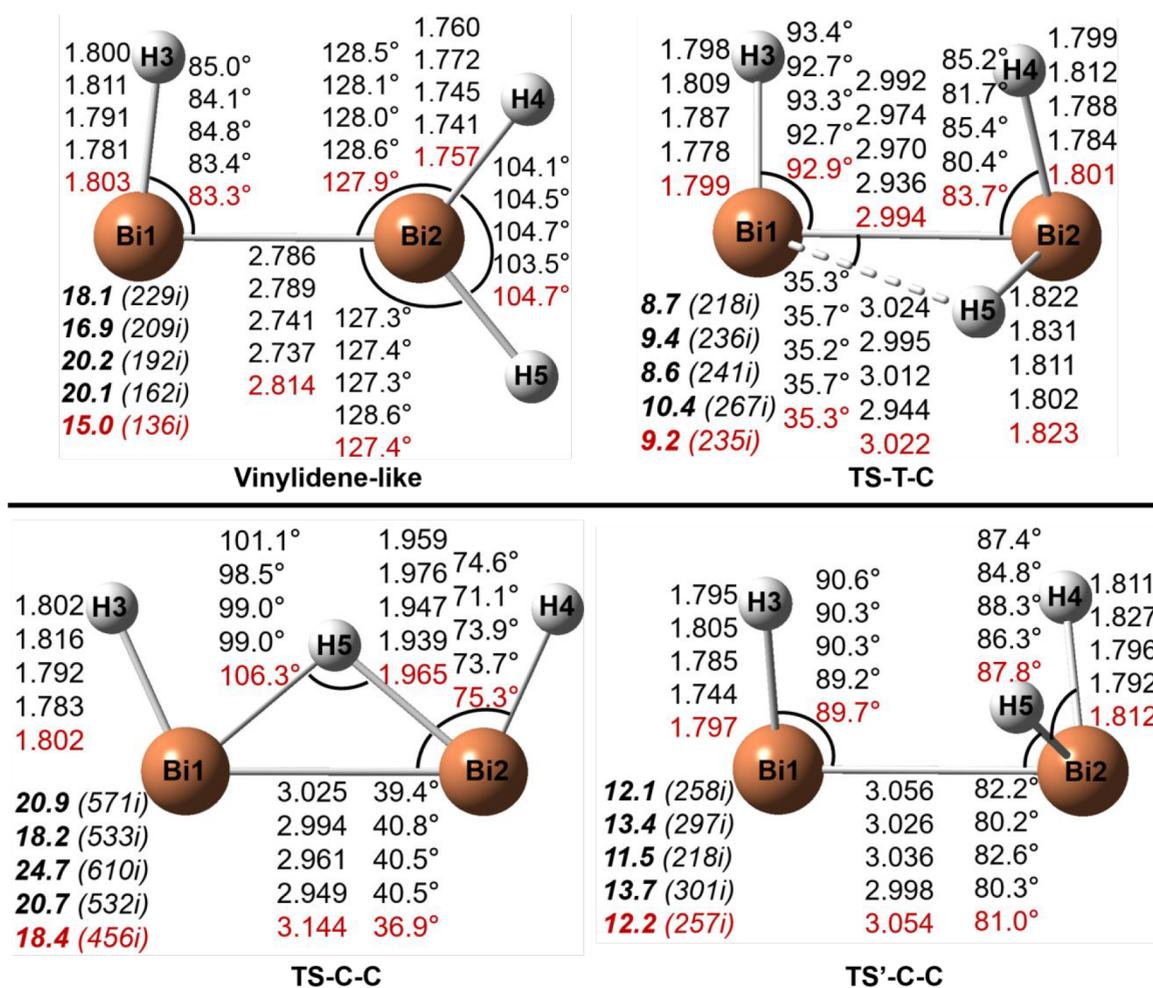


Figure 6. Geometries and energetics for four Bi_2H_3^+ transition states. Bond distances are in Å and energies (bold face) relative to the trans global minimum (in Figure 5) in kcal/mol. The imaginary vibrational frequencies (in cm^{-1}) are shown in parentheses.

Table 3. Relative Energies (ΔE , in kcal/mol) and Bi–Bi Bond Distances (R , in Å) for Bi_2H_3^+ at the CCSD(T)/cc-pVQZ-PP Level of Theory and Bi–Bi Wiberg Bond Indices (WBIs) and Bi Atomic Charges (Q_{Bi}) from Natural Bond Orbital (NBO) Analysis at the MN15/cc-pVTZ-PP Level

structure	ΔE	$R_{\text{Bi–Bi}}$	$\text{WBI}_{\text{Bi–Bi}}$	Q_{Bi}
trans	0.0	2.945	1.20	0.72/0.72
cis	1.6	2.964	1.18	0.71/0.71
vinylidene	15.0	2.814	1.71	0.67/0.41
TS-T-C	9.2	2.994	1.16	0.89/0.45
TS-C-C	18.4	3.144	1.31	0.64/0.64
TS'-C-C	12.2	3.054	1.04	0.93/0.41

the added proton (Figures 1, 3, and 5). The NBO analyses show that the electron distribution via a σ -hyperconjugation effect should occur between the two orbitals. Table 6 reports the second-order perturbation energies $E^{(2)}$ of the hyperconjugative $\sigma(\text{As}1\text{H}5) \rightarrow \text{LP}^*(\text{As}2)$ interactions, which are significant, i.e., 194 kcal/mol (As_2H_3^+), 153 kcal/mol (Sb_2H_3^+), and 142 kcal/mol (Bi_2H_3^+). These second-order perturbation values are comparable with the PA values reported above.

Observation by Microwave Spectroscopy. The rotational constants and dipole moments for the trans, cis, and vinylidene E_2H_3^+ structures obtained with different computa-

Table 4. Relative Energies (in kcal/mol) for the E_2H_3^+ Systems ($E = \text{As}, \text{Sb}, \text{Bi}$)^a

	As_2H_3^+	Sb_2H_3^+	Bi_2H_3^+
Cis	9.3	-0.4	-13.4
Trans	6.7	-2.4	-15.0
Vinylidene-like	0.0	0.0	0.0

^aGeometries were optimized with the CCSD(T)/cc-pVQZ-PP method.

tional methods are shown in Table S4. The results for the trans structures from different methods are close to each other. Unlike the E_2H_2 molecules, the trans structures for the E_2H_3^+ systems have dipole moments, which are 0.034 D (As_2H_3^+), 0.382 D (Sb_2H_3^+), and 0.461 D (Bi_2H_3^+) at the CCSD(T)/CC-PVQZ-PP level. The small dipole moment for trans As_2H_3^+ indicates that such a structure may be harder to observe in the microwave experiment. All cis E_2H_3^+ structures also have dipole moments, i.e., 1.186 D (As_2H_3^+), 0.640 D (Sb_2H_3^+), and 0.462 D (Bi_2H_3^+). The dipole moment of the vinylidene-like global minimum for As_2H_3^+ is 1.69 D. The large dipole moments of the vinylidene and cis As_2H_3^+ structures suggest that the microwave spectrum may be observable.

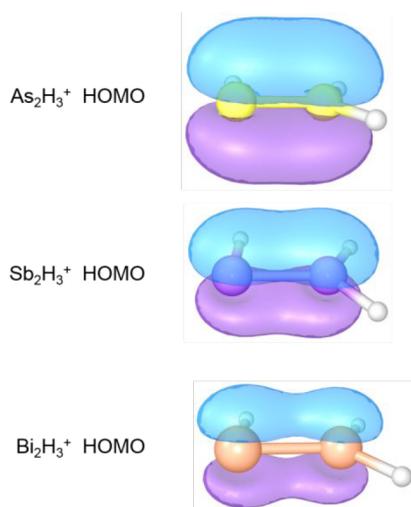


Figure 7. HOMO orbitals of the vinylidene-like structures of the As_2H_3^+ , Sb_2H_3^+ , and Bi_2H_3^+ systems.

Table 5. Proton Affinities (in kcal/mol) of E_2H_2 Molecules ($\text{E} = \text{As}, \text{Sb}, \text{Bi}$)^a

		As_2H_3^+	Sb_2H_3^+	Bi_2H_3^+
Cis		170.6	183.8	190.4
Trans		173.1	185.8	191.9
Vinylidene-like		179.8	183.4	177.0

^aAll three neutral E_2H_2 molecules have the trans structure lowest in energy. Thus, each proton affinity is computed with respect to the trans E_2H_2 structure. All geometries were optimized with the CCSD(T)/cc-pVQZ-PP method.

CONCLUSIONS

We have used four different DFT methods and the high-level CCSD(T) method to investigate the possible structures of the E_2H_3^+ ($\text{E} = \text{As}, \text{Sb}, \text{Bi}$) compounds. With substantial PAs (>180 kcal/mol), those protonated cations should be viable species. Overall, we report seven stationary points for As_2H_3^+ , eight for Sb_2H_3^+ , and six for Bi_2H_3^+ . For As_2H_3^+ , there are three minima (trans, cis, and vinylidene-like structures), among which the vinylidene isomer is the global minimum. For Sb_2H_3^+ , among the same three minima, the trans isomer is the lowest-lying structure. For Bi_2H_3^+ , the vinylidene isomer is a transition state, collapsing to the trans global minimum. The present theoretical work should be beneficial in future investigations of the E_2H_3^+ cations.

Table 6. Hyperconjugations and Second-Order Perturbation Energies $E^{(2)}$ (in kcal/mol) for Trans $\text{HE}=\text{EH}_2^+$ Structures ($\text{E} = \text{As}, \text{Sb}, \text{Bi}$) Predicted by NBO Analyses

$\text{HAs}=\text{AsH}_2^+$		$\text{HSb}=\text{SbH}_2^+$		$\text{HBi}=\text{BiH}_2^+$	
hyperconjugation	$E^{(2)}$	hyperconjugation	$E^{(2)}$	hyperconjugation	$E^{(2)}$
$\sigma(\text{As1H5}) \rightarrow \text{LP}^*(\text{As2})$	194.2	$\sigma(\text{Sb1H5}) \rightarrow \text{LP}^*(\text{Sb2})$	152.9	$\sigma(\text{Bi1H5}) \rightarrow \text{LP}^*(\text{Bi2})$	142.0
$\sigma(\text{As1H5}) \rightarrow \sigma^*(\text{As2H4})$	2.25	$\sigma(\text{Sb1H5}) \rightarrow \sigma^*(\text{Sb2H4})$	3.37	$\sigma(\text{Bi1H5}) \rightarrow \sigma^*(\text{Bi2H4})$	2.05
$\sigma(\text{As1As2}) \rightarrow \text{LP}^*(\text{As2})$	6.05	$\sigma(\text{Sb1Sb2}) \rightarrow \text{LP}^*(\text{Sb2})$	4.77	$\sigma(\text{Bi1Bi2}) \rightarrow \text{LP}^*(\text{Bi2})$	4.02

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.jpca.3c05945>.

Cartesian coordinates of optimized structures, harmonic vibrational frequencies, infrared intensities, dipole moments, and rotational constants for the E_2H_3^+ structures (PDF)

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Notes

The authors declare no competing financial interest.

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